

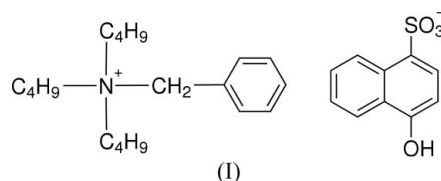
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## Key indicators

Single-crystal X-ray study  
 $T = 296$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å  
Some non-H atoms missing  
Disorder in main residue  
 $R$  factor = 0.049  
 $wR$  factor = 0.162  
Data-to-parameter ratio = 14.1For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.Benzyltributylammonium 4-hydroxy-  
naphthalene-1-sulfonateThe title compound,  $\text{C}_{19}\text{H}_{34}\text{N}^+\cdot\text{C}_{10}\text{H}_7\text{O}_4\text{S}^-$ , is a charge-control  
agent used in electrophotography. The anions form chains  
along the  $b$  axis through  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonding.Received 30 December 2006  
Accepted 9 April 2007

## Comment

The title compound, (I), is an ammonium salt used widely as a  
charge-control agent (CCA) of the positive type for toners in  
electrophotography (Tanaka, 1995). CCAs are usually added  
to toners to create a desired charge level and polarity (Nash *et al.*, 2001). However, the charge-control mechanism of CCA is  
not fully understood at the moment. We have, therefore,  
determined the title crystal structure as a step to elucidating  
the mechanism.Fig. 1 shows the asymmetric unit of (I). The ions have no  
crystallographically imposed symmetry. Fig. 2 shows a  
hydrogen-bonded chain along the  $b$  axis, formed by  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonding (Table 1).

## Experimental

Compound (I) was obtained from Orient Chemical Industries Ltd,  
and was recrystallized from a methanol solution. After 48 h, a  
number of colorless crystals were obtained in the form of blocks.

## Crystal data

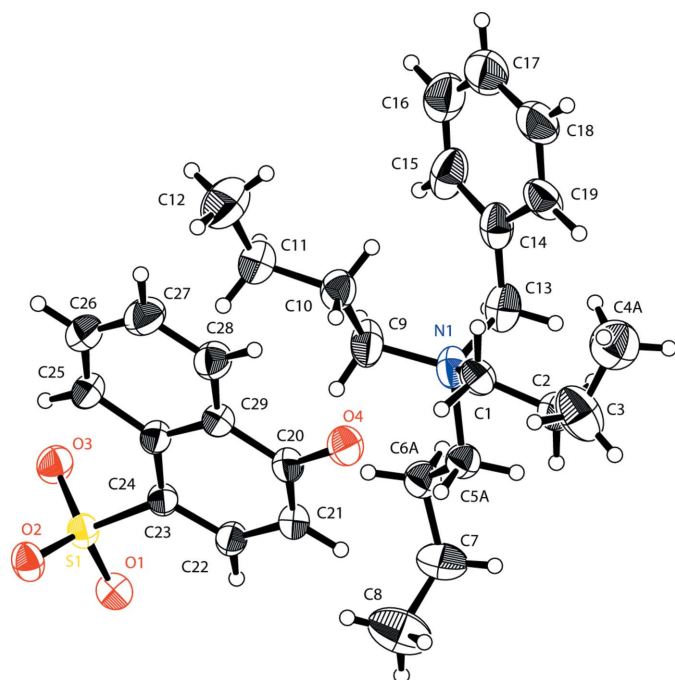
$\text{C}_{19}\text{H}_{34}\text{N}^+\cdot\text{C}_{10}\text{H}_7\text{O}_4\text{S}^-$	$V = 2787.8$ (4) Å <sup>3</sup>
$M_r = 499.70$	$Z = 4$
Monoclinic, $P2_1/n$	Cu $K\alpha$ radiation
$a = 14.3810$ (11) Å	$\mu = 1.29$ mm <sup>-1</sup>
$b = 9.8124$ (7) Å	$T = 296.1$ K
$c = 19.7757$ (15) Å	$0.20 \times 0.20 \times 0.20$ mm
$\beta = 92.560$ (5)°	

## Data collection

Rigaku R-Axis RAPID diffractometer	24226 measured reflections
Absorption correction: multi-scan ( <i>ABSCOR</i> ; Higashi, 1995)	4902 independent reflections
$T_{\min} = 0.745$ , $T_{\max} = 0.772$	3490 reflections with $F^2 > 2\sigma(F^2)$
	$R_{\text{int}} = 0.036$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.049$	347 parameters
$wR(F^2) = 0.162$	H-atom parameters constrained
$S = 1.19$	$\Delta\rho_{\max} = 0.45$ e Å <sup>-3</sup>
4902 reflections	$\Delta\rho_{\min} = -0.24$ e Å <sup>-3</sup>



**Figure 1**  
The asymmetric unit of (I), showing 30% probability displacement ellipsoids and only the major disorder components.

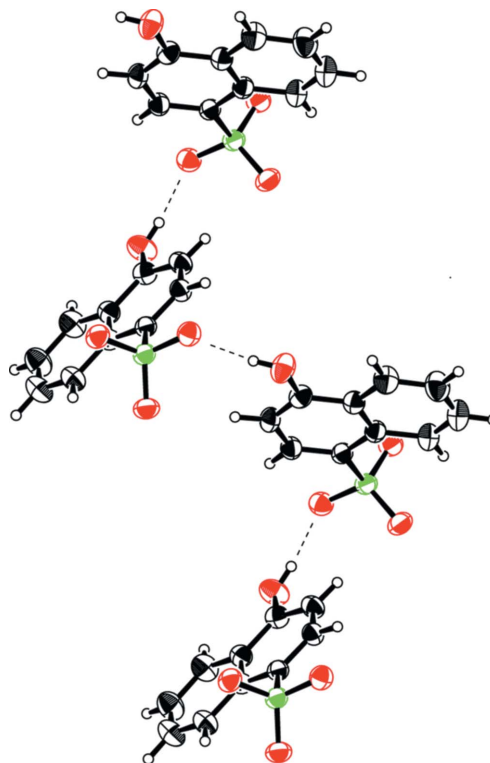
**Table 1**  
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O4-H4\cdots O1^i$	0.82	1.85	2.657 (2)	169

Symmetry code: (i)  $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$ .

C4, C5 and C6 were found to be disordered over two sites each. The site occupancies for C4A/C4B are 0.742 (7):0.258 (7), whereas those for C5A/C5B and C6A/C6B are 0.460 (6):0.540 (6). These atoms were isotropically refined. All H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C–H = 0.93 Å (aromatic), 0.96 Å (methyl) or 0.97 Å (methylene), and O–H = 0.82 Å;  $U_{iso}(H) = 1.2U_{eq}(\text{parent atom})$ .

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/ MSC, 2006); program(s) used to solve structure: *SIR2004* (Burla *et al.*,



**Figure 2**  
The formation of a hydrogen-bonded (dashed lines) chain.

2003); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP III* (Burnett & Johnson, 1996); software used to prepare material for publication: *CrystalStructure*.

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